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In this part of the worksheet, you will learn to

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easily classify compounds into one of the two categories. Table 1: Classification of Compounds ; Covalent Compounds Ionic Compounds ; N 2 O 4: AlF 3: CO 2: KNO 3: PCl 3: MgO : CH 4: Fe 2 O 3: NO 2: PbS : O 2: Na 3 N : In a covalent compound, valence electrons are shared between the two atoms in the bond. These can be evenly shared (covalent bond ...

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Complex Ion Nomenclature. The chemical formula and the chemical name for Complex cations have different rules. The complex cation chemical formula follows traditional rules for naming ionic compounds with the cation listed first. For example $[\text{Cu}(\text{OH})_2]_6^{2+}$, the cationic copper II ion is listed first and the neutral

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water ligand is listed second. Complex cation chemical names are assembled by ...

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Formulas and Nomenclature Binary Ionic Compounds

Worksheet Name the following compounds. 1. KCl 2. Li_2O 3. $CaBr_2$ 4. LiH 5. $MgBr_2$ 6. K_2O 7. ZnO 8. SrS 9. CaS 10. Ag_2S 11. ZnF_2 12. Ag_3N 13. NaF 14. BaO 15. Na_2S 16. $AlBr_3$ 17. Li_3N 18. KF 19. SrI_2 20. MgO 21. Al_2O_3 22. CaH_2 Write formulas. 1. sodium bromide 2. calcium oxide 3. silver chloride 4. silver

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oxide 5. aluminum ...

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Ionic Compounds Worksheet 1; Naming Ionic Compounds
Worksheet 2; Write Formulas for Ionic Compounds

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project you can use a worksheet. This worksheet provides extra practice for writing electron configurations. Write the unabbreviated electron configurations of the following elements.

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(b) Solubility of ionic compounds is usually governed by its lattice energy. In general the higher the lattice energy, the lower the solubility. The lattice energy of a salt gives a rough indication of the solubility of the salt in water because it reflects the energy needed to separate the positive and negative ions in a salt.

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